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of reactions and their mechanisms. Question.2. What do you understand by the rate of reaction ? Answer.

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To carry out the following reactions and classify them as physical or chemical changes. a. Iron with copper sulphate solution in water. b. Burning of magnesium in air. c. Zinc with dilute sulphuric acid d. Heating of copper sulphate e. Sodium

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chemistry involves a scientific approach to structure, applications of the organic molecules and compounds. Getting a job in the field of organic

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(i) Whether the reaction is fast, slow or instantaneous i.e. the rate of the reaction is not shown by a chemical equation. (ii) Whether the reaction goes to completion or is stopped in between i.e., the extent to which a reaction takes place is not known from a chemical equation. Q.11. Balance the following equations:

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Question 1. Write the balanced chemical equation with the

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state symbols of the following reaction: Solutions of Barium chloride and Sodium sulphate in water react to give insoluble Barium sulphate and the solution of Sodium chloride. (2017D) answer: $\text{BaCl}_2 (\text{aq}) + \text{Na}_2 \text{SO}_4 (\text{aq}) \rightarrow \text{BaSO}_4 (\text{s}) + 2 \text{NaCl} (\text{aq})$ Question 2.

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What is flow control? ... the reactant processed in the feed which means that reactants present in the feed transformed to products utilizing the chemical reaction and with the assistance of energy supplied to the system. 3. What is the solvent?

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equilibrium constant of chemical reaction _____ in the presence of catalyst. (A) Increases (B) Decreases ...

Chemical Reaction Engineering Objective Questions ans ...

The chemical equation showing the reaction between lime water and carbon dioxide is as follows: $\text{Ca(OH)}_2 + \text{CO}_2$

$\text{CaCO}_3 + \text{H}_2\text{O}$ CBSE Class 10 Science Previous Years' Solved Question Papers

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Viva Voce. Resources. Feedback. 1) Chemical reaction that absorbs heat from the surroundings is said to be _____ and has a _____ ΔH at constant pressure. Endothermic, negative. Exothermic, neutral. Endothermic, positive.

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exothermic, positive. 2)

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choice questions,lab viva 40 TOP CHEMICAL REACTION
Engineering Online Test ... Chemical reaction engineering
(reaction engineering or reactor engineering) is a specialty
in chemical engineering or industrial chemistry dealing with
chemical reactors. Frequently the term relates specifically to
catalytic reaction systems where either a ...

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What are some important Viva questions for chemistry
practical class 12 CBSE? - Quora. 1. Why is heating done for
Oxalic Acid ,KMnO₄ titration? 2. Why is H₂SO₄ used instead

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of HCl? 3. Explain the principle of Salt analysis.

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Chemical Reactions. VIVA VOCE. Question 1: Who gave the laws of chemical combination? Answer: Antoine Lavoisier and Joseph L. Proust. Question 2: Who coined the term atom? Answer: John Dalton. Question 3: Atoms are indivisible particles, which cannot be created or destroyed in a chemical reaction. Who said this? Answer: John Dalton. Question 4:

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Question: How can you predict how much energy is released in a chemical reaction? 1. Predict: In the warm-up activity, you observed how the reaction inside the chamber affected the temperature of the surrounding water. Based on what happens to the surrounding water, do you think heat energy (enthalpy) is absorbed in the reaction or released? Explain.

Question How can you predict how much energy is released

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Mole is the unit used to define the number of chemical substance present in a substance. It is the amount of substance which consists of the same number of chemical

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units as there are atoms in exactly 12 gram of pure carbon-12.

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VIVA NOTES (A) Volumetric Analysis (Titrations) Qs.1 What is Titration? Ans. The process of adding one solution from the burette into another in the conical flask in order to determine its volume after the completion of the chemical reaction is known as Titration. Qs.2 What is Neutralization?

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