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Mechanical Engineering Thermodynamics - Lec 27, pt 2 of 3: Example - First Law Gas Mixtures First Law of Thermodynamics, Basic Introduction, Physics Problems ~~Entropy Practice Problems, Enthalpy, Microstates, 2nd Law of Thermodynamics—Chemistry~~ Thermodynamics: Review of midterm exam, Maxwell relations (39 of 51) Thermodynamics: Overview of ideal gas mixtures, Amagat's and Dalton's laws (42 of 51) Thermodynamics Problems Solutions Cengel Boles

Substituting and multiplying by the factor 109 for the density unity kg/km³, the mass of the atmosphere is determined to be $m = 5.092 \times 10^{18}$ kg Discussion Performing the analysis with excel would yield exactly the same results. EES Solution for final result: $a=1.2025166$ $b=-0.10167$ $c=0.0022375$ $r=6377$ $h=25$ $m=4*\pi*(a*r^2*h+r*(2*a+b*r)*h^2/2+(a+2*b*r+c*r^2)*h^3/3+(b+2*c*r)*h^4/4+c*h^5/5)*1E+9$ 1-7 Pressure, Manometer, and Barometer 1-34C The pressure relative to the atmospheric pressure is called ...

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The results are: $(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$ for the unit of kg/m^3 , (or, $(z) = (1.20252 - 0.101674z + 0.0022375z^2) / 109$ for the unit of kg/km^3) where z is the vertical distance from the earth surface at sea level. At $z = 7$ km, the equation would give $\rho = 0.60 \text{ kg/m}^3$.

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Boles specializes in heat transfer and has been involved in the analytical and numerical solution of phase change and drying of porous media. He is a member of the American Society of Mechanical Engineers, the American Society for Engineering Education, and Sigma Xi. Dr. Boles received the ASEE Meriam/Wiley Distinguished Author Award in 1992 for excellence in authorship.

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